Impact of Wet and Dry Seasons on the Distribution of Polycyclic Aromatic Hydrocarbons in Selected Vended Street Foods in Parts of Port Harcourt Metropolis

G. I. Oyet¹, D. B. Kiin-Kabari¹*, M. O. Akusu¹ and S. C. Achinewhu¹

¹Department of Food Science and Technology, Rivers State University, Port Harcourt, P.M.B. 5080, Nigeria.

Authors’ contributions

This work was carried out in collaboration among all authors. Author GIO is a PhD student while other authors are his supervisors. The student managed the literature surfing of the study, performed the work and carried out statistical analysis. Authors DBKK and MOA designed the study. Author SCA wrote the protocol and wrote the first draft of the manuscript. All authors read and approved the final manuscript.

Article Information

DOI: 10.9734/EJNFS/2020/v12i130180

(1) Dr. Michael Lokuruka, Department of Food Science and Nutrition, School of Agriculture and Biotechnology, Karatina University, Karatina, Kenya.


(2) Krešimir Mastanjević, University of Osijek, Croatia.

Complete Peer review History: http://www.sdiarticle4.com/review-history/54483

Received 01 December 2019
Accepted 05 February 2020
Published 13 February 2020

ABSTRACT

The distribution patterns of PAHs in selected ready-to-eat street foods in parts of Port Harcourt metropolis was investigated during wet and dry seasons in 3 locations (Makoba-station 1, Elekahia-station 2 and Rivers State University-station 3). The study was carried out using a complete randomized design in three factorial experiments (Factors A, B and C). Factor A represented Season, B Location and C Street Vended foods samples. The selected food samples were Roasted plantain (RP1-3), Roasted Fish (RF1-3), Roasted Yam (RY1-3), Meat Pie (MP1-3), Suya (SY1-3) and Doughnut (DN1-3). The foods were sampled twice each season and the mean results recorded. Gas Chromatography-Mass Spectrophotometer (GC-MS) was used for the identification and evaluation of the presence of 16 Polycyclic Aromatic Hydrocarbon (PAHs).

*Corresponding author: Email: kabaridavid@yahoo.com;
Percentage distribution of PAHs in street vended foods during the wet and dry season showed naphthalene value of RY1:57.6% dry and RY1 Not Detected (ND) wet season, MP2: 10.7% dry and MP2: 3.4% wet. Higher naphthalene values distribution during dry season (DN1: 59.6%) was observed, with corresponding lower values recorded during the wet season (DN1: 43.3%). RP1: 10.4% wet and RY1: 19.4% wet while RP1: 9.6% dry and RY1: 2.6% dry showed lower percentage of Fluoranthene values during the dry season compared with higher values obtained for the wet season. Chrysene values (RP1: 10.9% wet, RP1: 10.0% dry, SY2: 69.2% wet, SY2: 71.4% dry, MP2: 69.8%, MP2: 22.7% wet) were detected in street vended food as low molecular weight hydrocarbons, with higher degree of distribution during dry season than the wet season. Higher molecular weight Benzo(a)anthracene was detected for all food samples. For RY2: 86.1% dry and 81% wet, RF3:71.3% dry and RF3: 52.0% wet, RF2: 69.0% wet, RF2: 61.4 dry, (DN1-DN3: 28-71.5% wet) and (DN1-3: 21.9-76% dry) seasons for Benzo(a)anthracene. The study showed that Benzo(a)anthracene had the highest percentage distribution during dry season in roasted fish and doughnut (DN2). Benzo(k)fluoranthene (RP1: 2.5% wet, 2.6%dry), Benzo(b)fluoranthene (R2: 9.9% wet, 1.7% dry, MP2: 8.9% dry and 2.7% wet) and Benzo(a)pyrene (RP1: 5.5% wet, 4.5% dry) were detected in all vended foods during wet and dry seasons, with higher percentage values observed during the dry season. Benzo(a)anthracene, Benzo(k)fluoranthene, Benzo(b)fluoranthene and Benzo(a)pyrene were detected in all vended foods. The study showed that the wet and dry seasons have imparted on the distribution levels of Lower Molecular Weight (LMW) and Higher Molecular Weight (HMW) of PAHs in ready-to-eat vended street foods. The patterns of distribution established the presence of these PAHs in selected ready-to-eat vended street foods. PAHs found in street vended foods is of public health concern to consumers and call for urgent attention for the review of the PAHs sources in food preparations, handling and storage in Port Harcourt metropolis.

Keywords: Wet and dry season; chromatography; PAHs; street vended foods; distribution patterns.

1. INTRODUCTION

Food is one of the prime necessities of life as water and air. Today, it is common knowledge that access to quality food is driven by the ability to pay. Most people are struggling to meet the food requirements of the family, hence have resorted to quick and cheaper foods vended along the street as a supplement, without recourse to good hygiene practices. Most workers such as drivers, mechanics, cleaners, operators, teachers, students and some middle-level employees, especially those working in urban business centers such as the universities and some areas in Port Harcourt metropolis, Rivers State have good patronage of street vended foods such as Akara, bread, roasted plantain (Bole), potatoes, yam, fish and corn. Some eat these foods as snacks on a daily basis. Also, some highly paid workers, have also found patronage of vended street food without recourse to food safety. This is also true of workers around oil and gas installations such as terminals and depots, truck marshaling yard, gas plants facilities along the Makoba areas and industrial road in Port Harcourt.

Street-foods are foods and beverages that are sold by street vendors or hawkers which could be raw or cooked [1]. Street vended foods are an extremely heterogeneous food category, encompassing meals, drinks and snacks. They show great variations in terms of ingredients, methods of retail, processing and consumption and are sold on the street from "pushcarts, baskets, balance poles, from stalls or shops having fewer than four permanent walls" [2].

In addition to food handling as contributors to food infection, the environment where these foods are prepared is a major source of concern. The seasonal variation, for instance during the dry season when the road is full of dust and other particulate matters from the environmental activities such heavy trucks movement will cause settlement of these dust particles on food samples and man as the eventual consumer become the receptor. At the petroleum depots, most mama-puts and other women roasting plantain, yam, potato, fish and frying of bean cake (Akara) are found around these facilities. They also served boiled or cooked foods without being sure of the preparation and handling processes as well as the storage temperature. In some cases, food such as stews are usually prepared from home and transported to the point of sales. In the raining season, we have also seen a deposit of soot’s arising from incessant burning by security agents of local boats and
containers used in the storage of locally refined petroleum products. Occupants of Port Harcourt metropolis enjoy wet and dry seasons all year round and therefore, faced the impact of the activities, products and services resulting from the significant environmental aspects of the oil and gas exploration. According to Umar et al. [3], the poor personal hygiene of the food vendors is a major source of public health concern. Polycyclic aromatic hydrocarbons (PAHs) are a group of notorious ubiquitous environmental contaminants produced primarily as a result of incomplete combustion of fossil fuels, biofuels and vegetation fires. They possess mutagenic, carcinogenic and teratogenic properties and are persistent in the environment having high lipophilicity [4]. PAHs studied in vended street food is of great concern due to their carcinogenicity, genotoxicity and mutagenicity properties [4]. Thus, the United States Environmental Protection Agency has listed 16 PAHs as priority pollutants on the basis of their occurrence and carcinogenicity. Polycyclic aromatic hydrocarbons are widespread in foodstuffs as a result of environmental pollution and some thermal treatments, which are used during the manufacturing of foods. Contamination of vended street foods by PAHs are known to be associated with various food processing procedures such as smoking, roasting, baking or frying [5]. Street food vending represents an important food security strategy for the low income population worldwide. However, no comprehensive risk analysis framework yet exists as regards to specific aspects of chemical/toxicological hazards in street foods. Indeed, all steps of street food production and vending can be vulnerable; from the selection of raw materials, through storage and preparation of meals and even the vending site are often exposed to urban pollutants. Relevant examples are cheap ingredients with illegal or undesirable residues, substances arising in poorly stored commodities (e.g. mycotoxins, histamine in scombroid fish) and metals leaching from cook-ware and process contaminants such as polycyclic aromatic hydrocarbons (PAHs) and acrylamide [6]. The Presence of PAHs in street vended foods is of public health concerns, hence the study is designed to evaluate the impact of wet and dry seasons on the distribution patterns of PAHs in street vended foods in selected parts of Port-Harcourt metropolis.

2. MATERIALS AND METHODS

2.1 Study Area

Port Harcourt is the capital and largest city in River State, located between the latitudes of 4°46‘38.71” N and longitudes of 7°00‘48.24” E in the heart of Niger Delta. The study was conducted in selected parts of Port Harcourt metropolis, Rivers States along the following sampling locations:

(a) Makoba: Terminal and Deports (Housing Oil and Gas, Truck Park/slump environment)
(b) Rivers State University gate (Urban, Academic Environment)
(c) Elekahia (urban, defining Industrial and Residential Area).

2.2 Chemicals

All chemicals and reagents used were of analytical grade. Dichloromethane (LC grade), alumina and silica gels were obtained from BDH (Poole, UK) while n-hexane was obtained from Aldrich (USA). A PAH standard mixture (NIST, Baltimore, MD) containing the 16 priority PAHs was used in this study and partial sodium salt-graft-poly (ethylene oxide) was from Sigma-Aldrich, Munich, Germany.

2.3 Methods

2.3.1 Experimental design

Six food samples each were purchased from the 3 (three) locations in Port Harcourt metropolis for two different days respectively and were wrapped with an aluminum foil paper, transported in an iced cooler to Food Chemistry Laboratory in the Department of Food Science and Technology, Rivers State University, Port Harcourt, Nigeria same day for analysis. One Air filter paper unexposed as control and the whole study was done using complete randomised design in a factorial experiment. Three factorials were used (Factors A, B and C); factor A represented Season, B Location and C Street Vended food samples given as 2X3X3 factorials. The vended foods are as shown in Table 1.

2.3.2 Sample collection

A total of 18 (Eighteen) food samples consisting of roasted fish, roasted plantain, roasted yam, meat pie, suya and doughnut were purchased
from roadside food vendors and hawkers along the Rivers State University Gate, Makoba-Industrial settlement, Elekahia-Urban dwellers and 3 (three) filter papers, all in Port Harcourt city, Rivers State, Nigeria. Three (3) samples were collected along the three (3) different locations for two (2) days during each season. They were wrapped in an aluminum foil, placed in a cooler and taken to the laboratory from which sub-samples were obtained for the determination of PAHs. The choice of the samples was carefully made to reflect the most consumed street vended foods in Port-Harcourt. The samples were stored at 4°C prior to analysis.

2.3.3 Determination of polycyclic aromatic hydrocarbons (PAHs)

PAHs were determined by Gas Chromatography-Mass Spectroscopy (GS-MS) using the ASTM D7363 – 07 methods. Five grams of the subsamples were mixed with the same amount of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodium salt of drying material poly (acrylic acid), partial sodi

Table 1. Experimental Design: Season and food samples with processing methods for the experiment

<table>
<thead>
<tr>
<th>Season</th>
<th>Station 1</th>
<th>Station 2</th>
<th>Station 3</th>
</tr>
</thead>
<tbody>
<tr>
<td>Wet and dry season</td>
<td>RP1</td>
<td>RP2</td>
<td>RP3</td>
</tr>
<tr>
<td></td>
<td>RF1</td>
<td>RF2</td>
<td>RF3</td>
</tr>
<tr>
<td></td>
<td>RY1</td>
<td>RY2</td>
<td>RY3</td>
</tr>
<tr>
<td></td>
<td>SY1</td>
<td>SY2</td>
<td>SY3</td>
</tr>
<tr>
<td></td>
<td>MP1</td>
<td>MP2</td>
<td>MP3</td>
</tr>
<tr>
<td></td>
<td>DN1</td>
<td>DN2</td>
<td>DN3</td>
</tr>
<tr>
<td></td>
<td>P1</td>
<td>P2</td>
<td>P3</td>
</tr>
</tbody>
</table>

Key: RP1 = roasted plantain from Makoba, RP2 = roasted plantain from Elekahia, RP3 = roasted plantain from Rivers State University, RF1 = roasted fish from Makoba, RF2 = roasted fish from Elekahia, RF3 = roasted fish from Rivers State University, RY1 = roasted yam from Makoba, RY2 = roasted yam from Elekahia, RY3 = roasted yam from Rivers State University, SY1 = roasted suya from Makoba, SY2 = roasted suya from Elekahia, SY3 = roasted suya from Rivers State University, MP1 = baked meat pie from Makoba, MP2 = baked meat pie from Elekahia, MP3 = baked meat pie from Rivers State University, DN1 = fried doughnut from Makoba, DN2 = fried doughnut from Elekahia, DN3 = fried doughnut from Rivers State University,
The detection and quantification limits (LODs and LOQs) were evaluated on the basis of noise obtained from the analysis of blank samples (n=4). The LOD and LOQ were defined as the concentrations of the analyte that produced a signal-to-noise ratio of 3 and 10, respectively. The r² values for the calibration curves in the concentration range of 2–100 μg ml⁻¹ were in range of 0.9992 to 0.9998 while LOD and LOQ values for the PAH compounds were in the range of 0.03 to 0.21 μg kg⁻¹ and 0.1 to 0.7 μg kg⁻¹, respectively. The instrument was calibrated using the manufactures standard calibration solution (Accustandard containing Alkanes mix 0.6 mg/mL and Accustandard PAH solution mix0.2 mg/mL CH2Cl2: MeOH (1:1).

2.4 Data Analysis

The mean values were subjected to statistical calculations which were performed using IBM SPSS (Statistical Package for Social Sciences) version 21. The analysis was done with GS-MS obtained at Rofnel Energy Services Limited, located at Plot 2 Addison Close, Rumuagholu, Port Harcourt, Nigeria.

3. RESULTS AND DISCUSSION

The study was structured to examine the percentage distribution of Polycyclic Aromatic Hydrocarbon (PAH) in street vended foods during wet and dry seasons in selected parts of Port Harcourt metropolis.

3.1 Station 1: Makoba

Polycyclic aromatic hydrocarbons such as Naphthalene, Acenaphthylene, Acenaphthene, Fluorene, Phenanthrene, Anthracene, Dibenzo (a, h) anthracene and Benzo (g, h, l) perylene and Indole (1, 2, 3-cd) Pyrene was not detected in roasted plantain and roasted yam sourced from Makobar during the wet and dry season. However, RY1 during dry season contained some LMW PAHs such as Naphthalene (57.6%) and Acenaphthene (27.7%). In the study Benzo (a) anthracene (60% wet, 62.1% dry), Chryssene (10.9% wet, 10.0% dry), Pyrene (7.5% wet,7.7% dry) and Fluoranthene (10.4% wet, 9.6% dry) were found the highest in roasted plantain in Makoba of the Σ16 PAHs while Benzo (k) Fluoranthene (2.5% wet, 2.6% dry), Benzo (b) Fluoranthene (3.2% wet, 3.5% dry) and Benzo (a) Pyrene (5.5% wet, 4.5 dry) were detected at lower percentages in roasted Plantain in Makoba during wet and dry season respectively. The study showed that suya, meat pie and doughnut contained varying percentages of LMW and HMW of PAHs at wet and dry season. Though, higher percentages of PAHs were observed during the dry season.

According to the findings in this study at station 1(Makoba) the lower molecular weight (LMW) PAHs, those with 2-3 aromatic rings were not detected in roasted plantain and Yam during the wet and dry season. The non-detection of these PAHs may be due to the seasonal effect of the rainy season and the cold precipitation of the dry season common at Makoba. Rain may have acted as cleaning agents to reduce the deposits and concentrations of the PAHs on food samples. The findings validate the work of Skupinska et al. [8] that once PAHs are emitted to the atmosphere, weight influences the fate of the gaseous PAH mixtures. Heavier PAHs (more than four rings) tend to adsorb to particulate matter, while lighter PAHs (less than four rings) tend to remain gaseous until removed via precipitation. PAHs concentrations in water tend to be low (around 100 mg/l) due to their weak solubility. The weak solubility leads to accumulation in sediments and aquatic organisms. PAHs can be absorbed by plants and can accumulate in soil. From this study, it was observed that Benzo (a) anthracene ranked highest (62.1% dry, 60.0% wet) and Benzo (a) pyrene at 5.5% wet, 4.5% dry concentrations of the Σ16 PAHs in roasted plantain at Makoba observed at dry and wet season. Similarly, Roasted fish, Meat Pie (Benzo (a) pyrene:5.0% dry, 0.4% Suya and Doughnut contained varying concentrations of these HMW PAHs. These findings corroborate with the work of Anjali and Dipanjali [9] who implicated the presence of Benzo (a) anthracene, Indole (1, 2, 3 –cd) pyrene and Benzo (a) pyrene in food samples as category 2 carcinogens, mutagen and teratogen, which is of public health concerns. In a similar study by Eke-Ejiofor and Maxwell [10], Polycyclic Aromatic Hydrocarbon (PAH) was detected in the roasted plantain samples, in some parts of Port Harcourt metropolis with levels ranging from 0.003–0.015 mg/kg.

3.2 Station 2: Elekahia

Roasted Plantain in Elekahia (RP2) recorded Naphthalene (32.6% wet, 30.4% dry), Acenaphthene, (7.9% wet, 8.3% dry) as 2-3 aromatic rings, while Benzo (a) anthracene (49.1% wet, 51.0% dry), Benzo (b) Fluoranthene (3.2% wet, 3.3% dry) being the highest 4
3.3 Station 3: Rivers State University

The major PAHs implicated in the Roasted plantain at Rivers State University were Chrysene (22.3% wet, 1.4% dry) and Benzo (a) anthracene (74.6% wet, 31.7% dry) 4-rings, respectively. Roasted plantain in Rivers State University did not show the presence of PAHs having 5 to 6 aromatic rings such as Dibenzo (a, h) anthracene, Indenol (1, 2, 3–cd) pyrene (trace amount: 0.2%) and Benzo (g, h, l,) perylene during wet and dry season respectively. However, the presence of Chrysene, Benzo (a) anthracene, Benzo (b) Flouranthene, Benzo (k) Flouranthene and Benzo(a)pyrene were observed in the street vended food samples at Rivers State University during wet and dry season respectively. The study also revealed that the Roasted plantain did not show the presence of Fluorene and Phenanthrene. The major PAHs of concern in the Roasted plantain at Rivers State University were Chrysene, Benzo (a) anthracene RY3 (76.7% wet season) 4-rings respectively. However, the presence of Chrysene, Benzo (a) anthracene, Benzo (b) Flouranthene, Benzo (k) Flouranthene and Benzo (a) pyrene were observed in small amount in the street vended food samples at wet season and dry season respectively. However, roasted yam at station 3 contained Naphthalene (0.7% wet, 50.0% dry), Acenaphthylene (0.2% wet, 0.7% dry), and Acenaphthene (0.1% wet, 30.6% dry) in low level during wet and higher values at dry season.

Roasted yam at wet and dry season when compared to roasted plantain, also contained the PAHs with 5 rings such as Benzo (k) Flouranthene (0.1% wet, 6.8% dry) and Benzo (b) Flouranthene (0.5% dry, 0.8% wet) which were observed at various percentages in roasted yam at Station 3 during wet season and dry season in small amounts, but capable of health risk to the consumer.

Roasted Fish (RF3) was found to contain all the Low Molecular Weight (LMW) 2 to 3 ringed PAHs such as Naphthalene (2.3% wet, 2.8%dry), during wet and dry season, with higher values found during the dry season. Similarly, High molecular weight, (HMW) 4 rings PAHs were detected in roasted fish (RF3), Chrysene, (23.3% wet, 42.7% dry), Chrysene for Suya, Meat Pie and Doughnut (SY3: 22.2% wet, 4.6% dry, MP3: 21.3% wet, 43.1% dry, DN3: 21.5% wet, 3.8% dry) Benzo (a) anthracene in SY3: 70.4% wet, 67.2% dry, MP3: 69% wet, 48.6% dry and DN3: 71.5% wet, 1.9% dry). Roasted fish like roasted plantain also contained the PAHs with 5 to 6 rings such as Benzo (b) Flouranthene (1%)and benzo (k) Flouranthene (0.2%) were detected at various percentages in roasted fish at station 3 during raining and dry season at no significant percentage difference. The presence of Benzo (a) anthracene, Benzo (a) pyrene and Benzo (b) Flouranthene, Dibenzo (a, h) anthracene, and Benzo (g, h, l,) perylene, and Indenol (1, 2, 3 – cd) pyrene in roasted fish, Suya, Meat Pie and...
doughnut during wet and dry season showed that PAHs is of public health concerns to the consumers of these food products. The presence of both the LMW and HMW of PAHs implicated the consumption of roasted fish, yam, plantain and doughnut as a regular meal. Similar findings by García et al. [12]; Alonge [11] revealed that PAHs with higher molecular weight (HMW) are more carcinogenic, mutagenic and teratogenic than the lower molecular weight (LMW) PAHs. Also, Anjali and Dipanjali [9] implicated all the 4-6 ringed PAHs as category 2 carcinogen, with benzo (a) pyrene characterized as not only category 2 carcinogen but mutagenic and teratogenic. The HMW PAHs do not show any preference for the seasonal variations.

The findings in this study corroborates the work of Lijinsky [13], Fritz and Soos [14]; Borokovcova et al. [15] and Emerole [16] who carried out a study to identify the occurrence of PAH in some Nigerian made local foodstuffs. They observed that significant values of benzo (a) anthracene and benzo (a) pyrene were detected in three varieties of smoked fish and smoked meat (suya) purchased from a popular market in Ibadan, Nigeria. Roasted fish may be implicated with the presence of percentages of LMW and HMW PAHs as detected in the present study due to mode of preparation such as smoking on a charcoal fire and other fire woods as the source of heat. This discovery is in agreement with the finding of De Vos et al. [17] that PAHs are discovered in significant amounts in some foods as a result of food processing methods such as cooking, smoking, preservation and storage, which are detected in meat, fish and vegetables.

According to Ziegler [18] who corroborates and implicated the food processing method as pathway for PAHs intake, he stated that “eating of a charcoal-broiled food may expose one to the same quantity of PAHs as one would receive from smoking 600 cigarettes”. Similarly, Bababunmi et al. [19], Fritz and Soos [14], Emerole [16] and Kazerouni et al. [20] documented through an epidemiological study that showed a statistical correlation between the increased occurrence of cancer of the intestinal tract and frequent intake of roasted food. From the findings in this study, one can state that people of Rivers State may be exposed to health risk as consumption of roasted fish and bole is ongoing unabated. This is being agreed by Alonge [11] who reported that PAHs are common and may constitute health hazards in Nigeria.

Recent studies have implicated the presence of HMW PAHs in the skin of roasted fish. Philips [21] reported that PAH concentrations in foodstuffs vary. Charring meat or barbecuing food over charcoal, wood, or other type of fire greatly increases the concentration of PAHs. For example, the PAH level for charring meat can be as high as 10–20 µg/kg.

In non-occupational settings, up to 70% of PAH exposure for a non-smoking person can be associated with diet [8]. Charbroiled and smoked meats and fish contain more PAHs than do uncooked products, with up to 2.0 µg/kg of benzo (a) pyrene detected in smoked fish. Tea, roasted peanuts, coffee, refined vegetable oil, cereals, spinach and many other foodstuffs contain PAHs. Some crops, such as wheat, rye and lentils may synthesize PAHs or absorb them via water, air or soil [22-25]. The current study reported increased levels of PAHs in fish and Amos-Tautua et al. [26] suggested that in order to reduce the high level of HMW PAHs in roasted fish which poses a high level of risk to consumers, that the burnt skin of roasted fish, meat or poultry should be avoided as it contained higher values of PAHs than the edible parts. From the current study, we discovered that the HMW carcinogenic PAHs constitute about 86% of the total PAHs in the roasted fish, of which 18% is accounted for by Chrysene and 65% by Benzo (a) anthracene, while BaP constituted only 3%. This was not in agreement with the findings of Amos-Tautua et al. [26] who reported that 33% of the total PAHs in the roasted fish, of which 7% is accounted for by BaP. More recently, Akpambang et al. [4] reported BaP at levels ranging from 2.4 to 31.2 µg/kg wet weights smoked fish and meat samples. Benzo (a) pyrene is the most studied PAHs and it is often used as a biomarker for PAHs.

It is important to note that benzo (a) pyrene was detected in traces amount (0.1 to 0.2%) in all 3 suya locations for this study and this is contrary to results obtained by Bababunmi et al. [19] who reported 8.5 µg/ kg of Benzo (a) pyrene (BaP) in suya meat. Duke and Albert [27] also found BaP contents ranging from 6.5 to 21.5 µg/ kg in suya meat from 4 different selling points. The result from the study showed that the average presence of benzo (a) anthracene (38.5%) and chrysene (53.3%) are at higher percentages when compared with the trend in fish. The present results showed a completely opposite value with the total PAHs being more in the suya...
than in the fish during the rainy season. This finding was in contrast to the work of Arpan et al. [28] that a strong correlation exists between fish lipids and PAH compounds; since PAH compounds are stored in fatty fish tissue. The HMW PAHs in fish and suya showed that PAHs are of food safety concern with its attendant carcinogenic tendencies and thus clearly make street vended suya unsafe for public consumption with its attendant health concerns to the consumers of suya across the various locations in Port Harcourt metropolis. The presence of these HMW PAHs may be attributed to the processing methods such as smoking in an open place. De Vos et al. [17], attributed the outdoor air pollution and the quality of water used in the cleaning of the meat before roasting or smoking. This agreed with the findings of Lioy et al. [29], obtained during a study of human exposure to Benzo (a) pyrene (BaP) that in some instances, outdoor air pollution led to a major portion of indoor air BaP exposures. Drinking water appeared to be a major pathway of BaP exposures in the study area. Benzo (b) Fluoranthene was detected in Meat Pie and doughnut (2.7 to 11.3% and 2.5 to 9.3%). Other HMW PAHs detected in traces amounts are Benzo (a) pyrene and Dibenzo (a, h) anthracene, Benzo (k) Fluoranthene and Benzo (g, h, i) perylene and Indeno (1, 2, 3–cd) pyrene found in Meat pie and Doughnuts. The finding of the presence of HMW PAHs in the current study corroborates with the work of Ahmed et al. [30] that meat pie had higher concentrations of the Σ16 PAHs compared to any other samples examined. The PAH profiles of the individual samples varied significantly, which may be associated with the heating temperature and fuel types used in the baking process [30]. Food is considered to be the major source of human PAH exposure due to PAH formation during cooking or from atmospheric deposition of PAHs on grains, fruits and vegetables. The relative contribution of airborne PAH pollutants to food levels (via fallout) has not been well characterized [17].

In a similar study by Chukwuindu [31] who stated that baked products were mainly contaminated with 2-, 3-, and 4-ring PAHs, which suggested that the contamination of these food items originated from automobile emissions, the combustion products of the fuel types used in the processing of these foods, baking procedure, and temperature. High Naphthalene value of 43.3%, 32.6% and 24.7% were found in the doughnut, Roasted Plantain and Meat pie from Makoba (DN1), (MP1) and Elekahia (RP2). The exposed Blank Paper also recorded a higher concentration of Chrysene PP1 (91.1%) Makoba, PP2 (88.8%) Elekahia and PP3 (67.1%) Rivers State University gate. A small increased risk of cancer in workers exposed to diesel exhaust as contributors of PAHs has been suggested by some epidemiologic studies [32]. The emergency of soots and daily burning of fossil fuels in Port Harcourt is of public health concerns to street vended food workers, traffic wardens, vehicle inspectors, drivers of trucks, mechanics etc. The high levels of PAHs detected in the street vended foods validates the work of Adonis et al. [33]; Kanoh et al. [34]; Kuo et al. [35] that higher levels of PAHs have been noted for residents of industrialized urban areas than in rural or suburban settings. Many-fold higher levels PAHs can be found in workers from certain occupations [36], including aluminum smelting [37]; diesel engine mechanics [33,38]; taxi, bus, and truck drivers [39,40]; painters [41], boiler makers [42]; toll booth operators [43]; traffic police [44] and coke oven plant workers [45-47].

EPA [48] has suggested that taking into your body each day “the following amounts of individual PAHs is not likely to cause any harmful health effects: 0.03 milligrams (mg) of anthracene, 0.06 mg of acenaphthene, 0.04 mg of fluoranthene, 0.04 mg of fluorene, and 0.03 mg of pyrene per kilogram (kg) of your body weight (one kilogram is equal to 2.2 pounds). Actual exposure for most of the United States population occurs from active or passive inhalation of the compounds in tobacco smoke, wood smoke, and contaminated air, and from eating the compounds in foods. Skin contact with contaminated water, soot, tar, and soil may also occur. Estimates for total exposure in the United States population have been listed as 3 mg/day.”

Seasonal variations, with much higher levels of DNA adducts in the wintertime, were observed both in residents of the district near the coke-oven area and in individuals from the rural area of Poland [49,50].

The OSHA-mandated PAH workroom air standard is an 8–hour time weighted average (TWA) permissible exposure limit (PEL) of 0.2 mg/m³, measured as the benzene-soluble fraction of coal tar pitch volatiles. The OSHA standard for coke oven emissions is 0.15 mg/m³. The National Institute for Occupational Safety and Health (NIOSH) has recommended that the
Table 2. Percentage distribution of polycyclic aromatic hydrocarbons (PAHs) in street vended foods during raining season

<table>
<thead>
<tr>
<th>PAH (%)</th>
<th>RP1</th>
<th>RP2</th>
<th>RP3</th>
<th>RY1</th>
<th>RY2</th>
<th>RY3</th>
<th>RF1</th>
<th>RF2</th>
<th>RF3</th>
<th>SY1</th>
<th>SY2</th>
<th>SY3</th>
<th>MP1</th>
<th>MP2</th>
<th>MP3</th>
<th>DN1</th>
<th>DN2</th>
<th>DN3</th>
<th>PP1</th>
<th>PP2</th>
<th>PP3</th>
</tr>
</thead>
<tbody>
<tr>
<td>Naphthalene</td>
<td>ND 32.6</td>
<td>0.9</td>
<td>ND 0.7</td>
<td>5.0</td>
<td>5.6</td>
<td>2.3</td>
<td>1.6</td>
<td>8.1</td>
<td>1.7</td>
<td>24.7</td>
<td>3.4</td>
<td>2.7</td>
<td>43.3</td>
<td>3.5</td>
<td>0.8</td>
<td>0.8</td>
<td>6.3</td>
<td>6.0</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Acenaphthyene</td>
<td>ND 0.5</td>
<td>0.1</td>
<td>ND 0.2</td>
<td>2.7</td>
<td>0.1</td>
<td>0.2</td>
<td>0.9</td>
<td>0.4</td>
<td>0.2</td>
<td>1.4</td>
<td>0.0</td>
<td>0.2</td>
<td>0.6</td>
<td>0.1</td>
<td>0.0</td>
<td>0.9</td>
<td>0.5</td>
<td>ND</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Acenapthene</td>
<td>ND 7.9</td>
<td>0.2</td>
<td>ND 0.1</td>
<td>2.9</td>
<td>1.2</td>
<td>0.8</td>
<td>0.3</td>
<td>1.7</td>
<td>1.2</td>
<td>6.4</td>
<td>1.6</td>
<td>1.0</td>
<td>10.0</td>
<td>0.6</td>
<td>0.5</td>
<td>ND</td>
<td>0.6</td>
<td>ND</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fluorene</td>
<td>ND 0.5</td>
<td>ND 15.6</td>
<td>ND 8.6</td>
<td>0.1</td>
<td>0.1</td>
<td>0.2</td>
<td>0.1</td>
<td>0.3</td>
<td>0.5</td>
<td>0.1</td>
<td>0.3</td>
<td>0.5</td>
<td>0.1</td>
<td>0.1</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>0.6</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Phenentrene</td>
<td>ND 0.5</td>
<td>ND 0.1</td>
<td>ND 0.1</td>
<td>0.1</td>
<td>0.1</td>
<td>ND</td>
<td>0.1</td>
<td>0.8</td>
<td>0.2</td>
<td>0.1</td>
<td>0.5</td>
<td>ND</td>
<td>0.1</td>
<td>0.8</td>
<td>0.8</td>
<td>1.3</td>
<td>ND</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Antracene</td>
<td>ND 0.5</td>
<td>ND 0.4</td>
<td>ND 0.3</td>
<td>0.2</td>
<td>0.1</td>
<td>0.9</td>
<td>0.5</td>
<td>1.0</td>
<td>ND</td>
<td>0.1</td>
<td>0.3</td>
<td>ND</td>
<td>1.0</td>
<td>0.8</td>
<td>ND</td>
<td>ND</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fluoranthene</td>
<td>10.4</td>
<td>1.3</td>
<td>19.4</td>
<td>1.8</td>
<td>0.2</td>
<td>5.9</td>
<td>0.5</td>
<td>0.1</td>
<td>0.3</td>
<td>0.2</td>
<td>0.2</td>
<td>0.9</td>
<td>0.2</td>
<td>0.1</td>
<td>1.4</td>
<td>0.1</td>
<td>0.2</td>
<td>ND</td>
<td>ND</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Pyrene</td>
<td>7.5</td>
<td>0.9</td>
<td>10.4</td>
<td>1.2</td>
<td>0.2</td>
<td>5.0</td>
<td>0.4</td>
<td>0.3</td>
<td>0.0</td>
<td>0.2</td>
<td>0.1</td>
<td>0.7</td>
<td>0.1</td>
<td>0.1</td>
<td>1.0</td>
<td>0.3</td>
<td>0.1</td>
<td>0.2</td>
<td>0.6</td>
<td>0.1</td>
<td></td>
</tr>
<tr>
<td>Chrysene</td>
<td>10.9</td>
<td>1.6</td>
<td>22.3</td>
<td>17.6</td>
<td>3.2</td>
<td>20.7</td>
<td>6.8</td>
<td>21.7</td>
<td>23.3</td>
<td>68.6</td>
<td>69.2</td>
<td>22.2</td>
<td>2.4</td>
<td>22.7</td>
<td>21.3</td>
<td>2.4</td>
<td>21.7</td>
<td>21.5</td>
<td>91.1</td>
<td>88.8</td>
<td>67.1</td>
</tr>
<tr>
<td>Benzo (a) antracene</td>
<td>60.0</td>
<td>49.1</td>
<td>74.6</td>
<td>16.8</td>
<td>81.0</td>
<td>76.6</td>
<td>54.7</td>
<td>69.0</td>
<td>52.0</td>
<td>26.6</td>
<td>18.7</td>
<td>70.4</td>
<td>48.0</td>
<td>68.0</td>
<td>69.0</td>
<td>28</td>
<td>70.8</td>
<td>71.5</td>
<td>1.9</td>
<td>0.5</td>
<td>11.6</td>
</tr>
<tr>
<td>Benzo (k) Fluoranthene</td>
<td>2.5</td>
<td>1.3</td>
<td>0.2</td>
<td>4.2</td>
<td>1.8</td>
<td>0.1</td>
<td>1.6</td>
<td>0.1</td>
<td>0.2</td>
<td>0.1</td>
<td>0.1</td>
<td>1.8</td>
<td>0.4</td>
<td>0.6</td>
<td>1.4</td>
<td>0.3</td>
<td>0.5</td>
<td>0.9</td>
<td>0.5</td>
<td>ND</td>
<td></td>
</tr>
<tr>
<td>Benzo (b) Fluoranthene</td>
<td>3.2</td>
<td>3.2</td>
<td>1.0</td>
<td>5.0</td>
<td>9.9</td>
<td>0.8</td>
<td>1.7</td>
<td>0.8</td>
<td>1.0</td>
<td>0.6</td>
<td>2.0</td>
<td>11.3</td>
<td>2.7</td>
<td>3.7</td>
<td>9.3</td>
<td>2.5</td>
<td>2.7</td>
<td>0.9</td>
<td>0.9</td>
<td>6.3</td>
<td></td>
</tr>
<tr>
<td>Benzo (a) Pyrene</td>
<td>5.5</td>
<td>0.6</td>
<td>11.0</td>
<td>1.0</td>
<td>0.1</td>
<td>2.8</td>
<td>0.1</td>
<td>0.1</td>
<td>0.2</td>
<td>0.1</td>
<td>0.4</td>
<td>0.1</td>
<td>0.2</td>
<td>0.8</td>
<td>0.0</td>
<td>0.1</td>
<td>ND</td>
<td>0.4</td>
<td>ND</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Dibenz (a, h) anthracene</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
<td>0.0</td>
<td>0.1</td>
<td>0.0</td>
<td>0.0</td>
<td>0.1</td>
<td>ND</td>
<td>0.2</td>
<td>0.1</td>
<td>ND</td>
<td>0.0</td>
<td>0.2</td>
<td>0.8</td>
<td>ND</td>
<td>ND</td>
</tr>
<tr>
<td>Indole (123-cd)</td>
<td>ND</td>
<td>0.0</td>
<td>ND</td>
<td>ND</td>
<td>0.1</td>
<td>ND</td>
<td>1.9</td>
<td>0.1</td>
<td>0.0</td>
<td>0.2</td>
<td>0.1</td>
<td>ND</td>
<td>0.6</td>
<td>ND</td>
<td>0.1</td>
<td>0.8</td>
<td>ND</td>
<td>0.0</td>
<td>ND</td>
<td>ND</td>
<td></td>
</tr>
<tr>
<td>Pyrene</td>
<td>ND</td>
<td>ND</td>
<td>0.0</td>
<td>ND</td>
<td>0.0</td>
<td>0.4</td>
<td>ND</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.3</td>
<td>0.1</td>
<td>0.1</td>
<td>0.2</td>
<td>ND</td>
<td>0.0</td>
<td>0.7</td>
<td>0.9</td>
<td>0.1</td>
<td>7.0</td>
<td></td>
</tr>
</tbody>
</table>
Table 3. Percentage distribution of polycyclic aromatic hydrocarbons (PAHs) in vended street foods during dry season

<table>
<thead>
<tr>
<th>PAH (%)</th>
<th>RP1</th>
<th>RP2</th>
<th>RP3</th>
<th>RP4</th>
<th>RP5</th>
<th>RP6</th>
<th>RP7</th>
<th>RP8</th>
<th>RP9</th>
<th>RP10</th>
<th>RP11</th>
<th>RP12</th>
<th>RP13</th>
<th>RP14</th>
<th>RP15</th>
</tr>
</thead>
<tbody>
<tr>
<td>Naphthalene</td>
<td>0.0</td>
<td>30.4</td>
<td>19.3</td>
<td>57.6</td>
<td>0.0</td>
<td>50.0</td>
<td>4.8</td>
<td>4.2</td>
<td>2.8</td>
<td>57.5</td>
<td>2.8</td>
<td>14.4</td>
<td>10.7</td>
<td>1.4</td>
<td>59.6</td>
</tr>
<tr>
<td>Acenaphthylene</td>
<td>0.0</td>
<td>0.6</td>
<td>0.6</td>
<td>1.4</td>
<td>0.0</td>
<td>0.7</td>
<td>2.4</td>
<td>0.1</td>
<td>0.3</td>
<td>1.1</td>
<td>0.1</td>
<td>0.0</td>
<td>1.4</td>
<td>0.3</td>
<td>0.4</td>
</tr>
<tr>
<td>Acenaphthene</td>
<td>0.0</td>
<td>8.3</td>
<td>6.5</td>
<td>27.7</td>
<td>0.0</td>
<td>30.6</td>
<td>2.8</td>
<td>0.9</td>
<td>1.2</td>
<td>27.2</td>
<td>1.3</td>
<td>0.3</td>
<td>6.8</td>
<td>5.1</td>
<td>1.9</td>
</tr>
<tr>
<td>Fluorene</td>
<td>0.0</td>
<td>0.5</td>
<td>0.2</td>
<td>2.1</td>
<td>1.1</td>
<td>2.8</td>
<td>7.3</td>
<td>0.1</td>
<td>0.1</td>
<td>2.1</td>
<td>0.1</td>
<td>0.3</td>
<td>0.5</td>
<td>0.3</td>
<td>0.2</td>
</tr>
<tr>
<td>Phenanthrene</td>
<td>0.0</td>
<td>0.4</td>
<td>1.5</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>5.6</td>
<td>0.1</td>
<td>0.1</td>
<td>0.0</td>
<td>0.1</td>
<td>9.5</td>
<td>0.8</td>
<td>0.5</td>
<td>0.4</td>
</tr>
<tr>
<td>Antracene</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.2</td>
<td>0.2</td>
<td>0.0</td>
<td>1.0</td>
<td>2.5</td>
<td>0.0</td>
<td>0.4</td>
<td>0.2</td>
</tr>
<tr>
<td>Fluoranthene</td>
<td>9.6</td>
<td>1.3</td>
<td>0.8</td>
<td>2.6</td>
<td>1.8</td>
<td>2.6</td>
<td>5.6</td>
<td>0.3</td>
<td>0.4</td>
<td>2.6</td>
<td>0.1</td>
<td>0.3</td>
<td>0.9</td>
<td>0.8</td>
<td>0.8</td>
</tr>
<tr>
<td>Pyrene</td>
<td>7.7</td>
<td>0.9</td>
<td>1.0</td>
<td>1.4</td>
<td>1.2</td>
<td>0.8</td>
<td>4.8</td>
<td>0.2</td>
<td>0.1</td>
<td>1.4</td>
<td>0.1</td>
<td>0.1</td>
<td>0.7</td>
<td>0.2</td>
<td>0.0</td>
</tr>
<tr>
<td>Benzo (a)</td>
<td>62.1</td>
<td>51.0</td>
<td>31.7</td>
<td>2.4</td>
<td>86.1</td>
<td>2.9</td>
<td>52.6</td>
<td>61.4</td>
<td>71.3</td>
<td>2.2</td>
<td>22.5</td>
<td>67.2</td>
<td>50.8</td>
<td>0.3</td>
<td>48.6</td>
</tr>
<tr>
<td>anthracene</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Chrysene</td>
<td>10.0</td>
<td>1.4</td>
<td>1.4</td>
<td>2.2</td>
<td>3.2</td>
<td>1.8</td>
<td>6.5</td>
<td>31.6</td>
<td>42.7</td>
<td>2.3</td>
<td>71.4</td>
<td>4.6</td>
<td>2.5</td>
<td>69.8</td>
<td>43.1</td>
</tr>
<tr>
<td>Benzo (b)</td>
<td>3.5</td>
<td>3.3</td>
<td>3.2</td>
<td>0.7</td>
<td>1.7</td>
<td>0.5</td>
<td>1.5</td>
<td>0.0</td>
<td>0.0</td>
<td>0.7</td>
<td>0.0</td>
<td>0.0</td>
<td>12.8</td>
<td>8.9</td>
<td>1.8</td>
</tr>
<tr>
<td>Fluoranthene</td>
<td>2.6</td>
<td>1.3</td>
<td>1.4</td>
<td>0.6</td>
<td>9.8</td>
<td>6.8</td>
<td>1.6</td>
<td>6.6</td>
<td>0.6</td>
<td>0.0</td>
<td>0.7</td>
<td>0.0</td>
<td>2.0</td>
<td>1.3</td>
<td>0.2</td>
</tr>
<tr>
<td>Benzo (k)</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Fluoranthene</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Benzo (a)</td>
<td>4.5</td>
<td>0.6</td>
<td>0.7</td>
<td>1.5</td>
<td>1.0</td>
<td>0.6</td>
<td>2.7</td>
<td>0.1</td>
<td>0.0</td>
<td>1.0</td>
<td>0.0</td>
<td>4.0</td>
<td>0.3</td>
<td>0.4</td>
<td>0.9</td>
</tr>
<tr>
<td>Pyrene</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Indole (123-cd)</td>
<td>0.0</td>
<td>0.2</td>
<td>0.2</td>
<td>0.0</td>
<td>0.1</td>
<td>0.0</td>
<td>1.4</td>
<td>0.0</td>
<td>0.1</td>
<td>1.0</td>
<td>0.1</td>
<td>0.3</td>
<td>0.8</td>
<td>0.0</td>
<td>0.0</td>
</tr>
<tr>
<td>Pyrene</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Dibenzo (a, h)antracene</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.1</td>
<td>0.0</td>
<td>0.1</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.5</td>
<td>0.2</td>
</tr>
<tr>
<td>Benzo (g, h, i)perylene</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.0</td>
<td>0.3</td>
<td>0.0</td>
<td>0.0</td>
<td>0.1</td>
<td>0.3</td>
<td>0.5</td>
<td>0.5</td>
<td>0.4</td>
<td>0.2</td>
</tr>
</tbody>
</table>

Key: RP1 = roasted plantain from Makoba, RP2 = roasted plantain from Elekahia, RP3 = roasted plantain from Rivers State University, RF1 = roasted fish from Makoba, RF2 = roasted fish from Elekahia, RF3 = roasted fish from Rivers State University, RY1 = roasted yam from Makoba, RY2 = roasted yam from Elekahia, RY3 = roasted yam from Rivers State University, SY1 = roasted suya from Makoba, SY2 = roasted suya from Elekahia, SY3 = roasted suya from Rivers State University, MP1 = baked meat pie from Makoba, MP2 = baked meat pie from Elekahia, MP3 = baked meat pie from Rivers State University, DN1 = fried doughnut from Makoba, DN2 = fried doughnut from Elekahia, DN3 = fried doughnut from Rivers State University.
workplace exposure limit for PAHs be set at the lowest detectable concentration, which was 0.1 mg/m³ for coal tar pitch volatile agents at the time of the recommendation [51].

Redmond et al. [52] reported increased incidences of lung, skin, and bladder cancers that are associated with occupational exposure to PAHs. Epidemiologic reports of PAH exposed workers have noted increased incidences of skin, lung, bladder, and gastrointestinal cancers. These reports, however, provide only qualitative evidence of the carcinogenic potential of PAHs in humans because of the presence of multiple PAH compounds and other suspected carcinogens. Some of these reports also indicate the lack of quantitative monitoring data [53-55,52].

Later experimental studies showed that PAHs in soot were probably responsible for the increased incidence of scrotal cancer noted by Percival Pott among London chimney sweeps in his 1775 treatise [56].

4. CONCLUSION
The study was structured to examine the percentage distribution of Polycyclic Aromatic Hydrocarbon (PAH) in street vended foods during raining and dry seasons. According to the findings in this study at station 1 (Makoba) the lower molecular weight (LMW) PAHs, those with 2-3 aromatic rings such as Naphthalene, Acenaphthylene, Acenaphthene, Fluorene, Phenanthrene and Anthracene were not detected in roasted plantain and Fish. However, HMW PAHS (4-6 ringed) such as Benzo (a) anthracene, Chrysene, Pyrene and Fluoranthene were found the highest in roasted plantain in Makoba (60%, 10.9%, 10.4% and 7.5%). The presence of both the LMW and HMW of PAHs in SVF implicated the consumption of street vended food as a regular meal. The presence of Benzo (a) anthracene, Benzo (a) pyrene and Benzo (b) Fluoranthene, Dibenzo (a, h) anthracene, and Benzo (g, h, l) perylene, and Indeno (1, 2, 3-cd) pyrene in roasted fish, Doughnut, Suya and plantain showed that PAHs is of public health concerns to the consumers of roasted fish, yam and plantain popularly known as bole in Port Harcourt metropolis. This study has highlighted the existence of PAHs getting introduced into our food chain.

COMPETING INTERESTS
Authors have declared that no competing interests exist.

REFERENCES
2. FAO. Food and Agriculture Organization. Report on: Improving the nutritional quality of street foods to better meet the micronutrient needs of school children in Urban Areas. 2007;14-17.


29. Liou PL, Waldman JM, Greenberg A, Harkov R, Pietarinen C. The Total Human Environmental Exposure Study (THEES) to benzo(a)pyrene: Comparison of the inhalation and food pathways. Archives of
DOI: 10.1080/00039896.1988.10545954

DOI: 10.1016/s0378-1017(00)00300-9

DOI: 10.1556/AAlim.2015.0004


Available: https://doi.org/10.1016/S0378-4274(03)00225-X

DOI: 10.1006/enrs.1993.1108

DOI: 10.1016/j.jchromb.2003.12.012


DOI: 10.1007/s00420-003-0477-y


Available: https://doi.org/10.1021/es030588k


DOI: 10.1080/00039890209602945

46. Serdar B, Waidyanatha S, Zheng Y, Rappaport SM. Simultaneous determination of urinary 1- and 2-napththols, 3- and 9-phenanthrols, and 1-pyrenol in coke oven workers. Biomarkers: Biochemical Indicators of Exposure,
DOI: 10.1080/1354750021000046570

DOI: 10.1136/oem.2002.006643


49. Grzybowska E. Seasonal variations in levels of DNA adducts and X-spots in human populations living in different parts of Poland. Environmental Health Perspectives. 1993;99:77-81.

50. Motykiewicz G. Application of biomarkers in heavily polluted industrialized areas of countries of Central and Eastern Europe. Toxicology. 1995;101:117-123.


DOI:https://doi.org/10.1080/00022470.1975.10470095